

Action of Phenobarbital Given to Rats Together with Diethylnitrosamine on the O⁶-Ethylguanine Content of Liver DNA

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Abstract—When rats are fed diethylnitrosamine (10 mg/kg/day), no O⁶-ethylguanine is found in liver DNA after 2 weeks, but a considerable amount accumulates after 4 weeks. On the other hand, a 2-week feeding of diethylnitrosamine is not sufficient to induce liver cancers, whereas a 4-week treatment leads to hepatocarcinomas in 50% of the animals.

Administration of phenobarbital (75 mg/kg/day) together with diethylnitrosamine during 4 weeks prevents the formation of liver cancers. It also prevents accumulation of O⁶-ethylguanine in liver DNA.

Phenobarbital does not change the amount of O⁶-ethylguanine repair activity present in liver chromatin after 2 or 4 weeks of treatment with diethylnitrosamine. It is thus concluded that, by inducing the development of the endoplasmic reticulum, phenobarbital decreases the equilibrium concentration of the ultimate carcinogen derived from this indirect alkylating agent.

INTRODUCTION

DIETHYLNITROSAMINE (DNA) is an indirect alkylating agent that, after enzymatic activation in liver, generates the carbocation CH₃-CH₂⁺ which reacts with nucleophiles in DNA. Several studies have suggested that alkylation of guanine O-6 can lead to mutation and cancer [1-4]. Barbason and Betz [5] have shown that continuous administration of DNA to rats (10 mg/kg/day) for 2 weeks induces preneoplastic foci, but is not sufficient to produce neoplastic nodules; these nodules appear only when the DNA administration has lasted for 4 weeks or more.

When phenobarbital (PhB) is given after the DNA administration, it shortens the latency period between the appearance of preneoplastic foci and their malignant transformation [6]; this promoter effect of PhB given after an initiator has been found in numerous studies [7-9].

By contrast, PhB given simultaneously with DNA reduces liver cancer frequency [10, 11]. Barbason *et al.* [12] observed that the number and size of preneoplastic lesions are lower when a 4-week DNA treatment is associated with the administration of PhB; moreover not a single cancer develops whereas, when DNA is given alone, 50%

of the rats show hepatocarcinomas at *post mortem* examination.

Since there might be a correlation between liver cancer and O⁶-alkylation of guanine, we have hypothesized that PhB given together with DNA would decrease the O⁶-ethylguanine (O⁶-etG) content of liver DNA. The O⁶-etG content of DNA depends on two parameters: (i) the concentration-time integral of CH₃-CH₂⁺ in the cell; (ii) the rate of O⁶-etG repair. Renard *et al.* [13] observed the disappearance of O⁶-etG from DNA in isolated rat liver nuclei. The repair mechanism is a transethylation from O⁶-etG onto a cysteine of a chromatin protein which can be used only once (suicide reaction) [14, 15].

In this work, we have measured the O⁶-etG content of liver DNA after administration to rats of DNA alone or together with PhB. We have also determined the amount of repair protein in liver chromatin in both animal groups.

MATERIALS AND METHODS

Wistar male rats of about 180 g received DNA and/or PhB in drinking water for 2 or 4 weeks. The daily doses were about 10 mg/kg for DNA, and 75 mg/kg for PhB. The animals were sacrificed after a 18-h fast.

Preparation of liver DNA

The livers were homogenized in 1 M NaCl, 30 mM *N*-lauroyl-sarcosine (5 ml/g liver). Water-saturated phenol, 70 mM 8-hydroxyquinoline, was added to the homogenate and the mixture shaken for 30 min at room temperature. After centrifugation, the aqueous phase containing DNA was recovered and submitted to another deproteinization with an equal volume of chloroform/isoamyl alcohol (24:1; v:v). The mixture was centrifuged and the upper phase collected. This solution was submitted to two successive digestions of 30 min at 37°C, the first with pancreatic ribonuclease (40 µg/g liver) and the second with proteinase K (200 µg/g liver). After a last treatment with chloroform/isoamyl alcohol, DNA was precipitated from the solution with an equal volume of ethoxyethanol, and the pellet redissolved in 0.15 M NaCl, 0.015 M Na citrate, pH 7.0. The DNA solution was dialyzed against 10 mM Tris-HCl, 1 mM EDTA, pH 8.0, then concentrated on polyethylene glycol 6000.

The amount of DNA per ml was determined by the diphenylamine method [16]. Samples containing 1.5 mg DNA were depurinated in 0.1 M HCl at 70°C for 50 min. After addition of 1 M K₂HPO₄ to reach pH 4.0 and lyophilization, 1 ml 50 mM ammonium phosphate, pH 4.0, was added. The samples were centrifuged 10 min at 15,000 *g* and the supernatants taken for HPLC analysis.

Purine analysis

The samples (250 µl) were chromatographed on an Altex ODS-10 ultrasphere column (25 × 1 cm) in an HPLC Altex 332. The elution was carried out at a rate of 2 ml/min with 50 mM ammonium phosphate, pH 4.0/acetonitrile (85:15; v:v). The O⁶-etG retention time was 11 min.

Preparation of chromatin proteins

Cell nuclei and chromatin were prepared from rat liver according to Thibodeau and Verly [17]; non-histone proteins were extracted from chromatin with heparin-ultragel as described by Renard and Verly [18]. This extract was dialyzed against 50 mM Tris-HCl, 1 mM EDTA, 2 mM dithiothreitol, pH 8.0. The amount of protein was determined by the method of Bradford [19].

Transalkylase assay

The enzyme activity was measured as described by Renard and Verly [18] and Lemaître *et al.* [20].

Calf thymus DNA was alkylated with [³H]ethyl-nitrosourea (4.5 Ci/mmol; IRE, Fleurus, Belgium), and stored at a concentration of 1 mg/ml in 50 mM Tris-HCl, 1 mM EDTA, 2 mM dithiothreitol, pH 8.0, at 4°C.

Three volumes of chromatin protein solution were incubated with one volume of [³H]ethylated DNA

solution for 60 min at 37°C. To 200-µl aliquots (50 µg DNA containing 285 fmol O⁶-etG; 25 µg protein) were added 5 µg non-radioactive O⁶-etG in 10 mM HCl, and 33 µl 1M HCl. The mixture was heated at 70°C for 50 min. After addition of 35 µl 1M K₂HPO₄ and 57 µl acetonitrile, and centrifugation, 250-µl samples were taken from the supernatant for HPLC analysis.

The O⁶-etG-containing fraction is collected after 11 min and its radioactivity measured. Knowing the chromatographic yield (from the non-radioactive O⁶-etG internal standard) and the specific radioactivity of the alkylating agent, the amount of O⁶-etG in the DNA submitted to depurination can be calculated.

RESULTS

Accumulation of O⁶-etG in liver DNA

Male rats of the same age were randomly distributed between two groups. Animals of the first group were treated for 2 weeks, and those of the second group for 4 weeks. Each group was divided in two: rats of one subgroup received DENA alone; rats of the other subgroup received DENA and PhB. At the end of the treatments and at regular intervals for 6 weeks afterwards, three rats were sacrificed in each subgroup. The livers were pooled and the DNA extracted; 1.5 mg DNA was depurinated and the purines separated by HPLC.

The amount of O⁶-etG in the DNA sample could be estimated by reference to a calibration curve obtained by injecting increasing amounts of synthetic O⁶-etG in the HPLC apparatus. Knowing that 24.5% of the bases in rat DNA are guanines, the O⁶-etG/10⁶ G ratio in the analyzed DNA could be calculated.

Liver DNA did not contain significant amounts of O⁶-etG after 2 weeks of DENA treatment alone, or associated with PhB. This means that the transalkylase removed the ethyl group from O⁶-etG as soon as this adduct appeared in the cell DNA.

After 4 weeks of treatment, the results are very different in the two subgroups. The rats that received DENA alone had accumulated 1600 O⁶-etG/10⁶ G in their DNA at the end of the treatment. This indicates that the repair capacity had been overridden. By contrast, the rats that received DENA and PhB had accumulated only 15 O⁶-etG/10⁶ G. After the end of the treatment, the O⁶-etG progressively disappeared from liver DNA in both subgroups (Table 1); this is due to a residual repair activity (see later) and also likely to the synthesis of new transalkylase molecules.

Repair activity

Two groups of rats, treated for 2 and 4 weeks respectively, were divided into subgroups. The animals of the first subgroup were kept as a control;

Table 1. Action of phenobarbital on the O⁶-ethylguanine content of liver DNA in rats treated with diethylnitrosamine

Treatment	Day			
	28	32	35	42
DENA	1600	50	5	1
DENA + PhB	15	3	1	0

Rats received *per os* DENA (10 mg/kg/day) with or without PhB (75 mg/kg/day) for 4 weeks. They were sacrificed in lots of three at the end of the treatment (day 28) or later. The O⁶-etG content of liver DNA was measured; the results are expressed as O⁶-etG/10⁶G ratios.

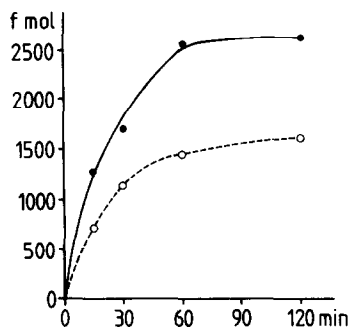


Fig. 1. In vitro repair of O⁶-ethylguanine in DNA by chromatin proteins. Chromatin non-histone proteins, prepared from livers of rats treated 2 weeks (●—●) or 4 weeks (○-○) with DENA, were incubated with DNA alkylated with ethylnitrosourea. The amount of O⁶-etG that disappeared from DNA was followed as a function of time; the results are expressed as f mol O⁶-etG repaired by the protein from an amount of chromatin containing 1 mg DNA.

the animals of the second received DENA alone; the animals of the third received PhB alone; the animals of the fourth received DENA and PhB. At the end of the treatment, rats and their livers were weighed; the non-histone proteins were extracted from the liver chromatin of the four rats constituting each subgroup.

Livers from rats receiving PhB were enlarged, less however when they were fed DENA simultaneously.

Three volumes of chromatin-protein solution (200 µg/ml) were incubated at 37°C with one volume of [³H]ethylated DNA (DNA alkylated with [³H]ethylnitrosourea) solution (1 mg/ml). Three 200-µl aliquots were taken after 15, 30, 60 and 120 min. After depurination and HPLC analysis, the radioactivity in the O⁶-etG peak was measured. Samples of [³H]ethylated DNA were incubated without chromatin proteins to measure the amount of O⁶-etG loss in absence of repair activity. The results, corrected for this control, enable to calculate the amount of O⁶-etG that disappeared from the [³H]ethylated DNA due to repair processes; the results are expressed in f mol O⁶-etG removed from the [³H]ethylated DNA by the chromatin proteins corresponding to 1 mg liver DNA.

Figure 1 shows that the amount of O⁶-etG repair

Table 2. Amounts of transalkylase in chromatin of rats fed diethylnitrosamine and/or phenobarbital

Treatment	fmol/mg liver DNA	
	2 weeks	4 weeks
Control	342	375
DENA	2570	1460
PhB	473	566
DENA + PhB	2226	1301

Rats were fed DENA and/or PhB for 2 or 4 weeks. Non-histone proteins were extracted from liver chromatin and incubated with DNA alkylated with ethylnitrosourea during 60 min at 37°C. The amount (fmol) of O⁶-etG that disappeared from DNA is equal to the amount of transalkylase. The results are given as fmol of transalkylase in a chromatin amount containing 1 mg DNA.

reached a plateau after 60 min. We recall that the repair of O⁶-etG is a suicide reaction: the amount of O⁶-etG that had disappeared when the reaction stopped (fmol) is equal to the amount of transalkylase in the medium. The results are presented in Table 2.

We shall first comment on the situation after 2 weeks of treatment. The residual repair activity in the chromatin of rats which were fed DENA is seven times higher than the activity found in the control rats. We have seen, in the previous section, that this increased activity was sufficient to mop up the O⁶-etG lesions as soon as they appeared. More pertinent to our subject is the observation that PhB administration did not change significantly the repair activity whether DENA was simultaneously given or not.

The situation after 4 weeks again shows that PhB feeding had no significant effect. On the other hand, the rats that received DENA had a repair activity higher than the controls, but which was however lower than that observed after 2 weeks, suggesting a progressive exhaustion of the repair capacity likely explaining the accumulation of O⁶-etG in liver DNA of rats fed DENA alone.

DISCUSSION

We show that, when 10 mg DENA/kg/day is continuously fed to rats for 2 weeks, no O⁶-etG can be found in liver DNA; the repair activity is sufficient to eliminate the adduct as soon as it is formed. This result correlates well with the observation that a 2-week DENA treatment is not sufficient to induce liver cancers [6].

When the DENA feeding is prolonged for 4 weeks, a large amount of O⁶-etG (1600/10⁶ G) has accumulated in liver DNA. Thus a 4-week DENA treatment seems to override the repair capacity so that some O⁶-etG is continuously present in liver DNA during the last days; it, however, disappears after the end of the treatment. On the other hand,

50% of the rats die with liver cancers after about 12 months [6].

When, during the 4 weeks, PhB (75 mg/kg/day) is given together with DENA, the amount of O⁶-etG found in liver DNA at the end of the treatment is 100-times smaller. The addition of PhB to DENA also prevents the appearance of liver cancers [12].

Because of the rapid disappearance of O⁶-etG from rat liver DNA, Scherer *et al.* [21] have argued that O²- and O⁴-ethylthymine, which are minor alkylation products but which persist longer in liver DNA, might be responsible for liver cancers due to DENA rather than O⁶-etG. Without denying a role for these O-alkylated pyrimidines, our results rather favor O⁶-etG as the main carcinogenic lesion since liver cancer seems to be induced only when the repair system is exhausted and O⁶-etG begins to accumulate in liver DNA.

To investigate the mechanism by which PhB prevents the accumulation of O⁶-etG in liver DNA between 2 and 4 weeks of DENA feeding, we have measured the repair activity of the chromatin proteins. We found that the DENA treatment increased the amount of transalkylase in liver chromatin, confirming results obtained by Margison *et al.* [22] and Renard *et al.* [23]. On the other hand, the PhB treatment did not change the repair activity, expressed per mg DNA, in liver chromatin (constitutive activity for rats not receiving DENA,

or induced activity for rats fed DENA).

If PhB does not change the repair activity, the lower amount of O⁶-etG found after a 4-week DENA feeding if PhB is simultaneously given can only be explained by an action of PhB on the concentration of the ultimate carcinogen derived from DENA in liver cells. This ultimate carcinogen is believed to be the carbocation CH₃-CH₂⁺; it is produced and destroyed by enzymes located in the endoplasmic reticulum. PhB induces an important development of the endoplasmic reticulum; one must suppose that the increase of the microsomal enzyme activities leads to a new dynamic equilibrium with a lower concentration of the ultimate carcinogen. The repair activity is then sufficient to eliminate rapidly the O⁶-etG that appears in liver DNA even when the DENA treatment lasts 4 weeks.

It must be underscored that the final interpretation of the PhB protective effect (a lower CH₃-CH₂⁺ concentration) explains the decreased incidence of cancers independently of the fact that the oncogenic lesion is O⁶-etG or another ethylated base.

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